# The Crystal Structure of $\mathbf{C u T e O}_{4}$ 

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Crystals of $\mathrm{CuTeO}_{4}$ are disordered. The structure has been determined in a unit cell with $a=5 \cdot 500(2), b=$ 10.327 (7), $c=4.704$ (2) $\AA, \alpha=\beta=\gamma=90.0^{\circ}, Z=4$. Structural details have been obtained from a refinement in space group $P 2_{1} / n(R=0.048)$, and the OD character of the structure is discussed in terms of idealized $\left[\mathrm{TeO}_{4}\right]_{n}^{2 n-}$ layers with the plane group symmetry $\operatorname{Pm}(n) a(R=0.054)$. The Cu atoms have a distorted octahedral coordination of O atoms with $\mathrm{Cu}-\mathrm{O}$ distances in the range $1.91-2.26 \AA$. The $\mathrm{Te}-\mathrm{O}$ coordination is also octahedral; the octahedra are connected by corners forming $\left[\mathrm{TeO}{ }_{4}\right]_{n}^{2 n-}$ layers. $\mathrm{The} \mathrm{Te}-\mathrm{O}$ bonds involving terminal and bridging O atoms are 1.87-1.91 and 1.91-2.02 $\AA$ respectively.

## Introduction

The structures of some $\mathrm{Cu}^{11}$ tellurates(IV) have been determined: $\mathrm{CuTeO}_{3} \cdot 2 \mathrm{H}_{2} \mathrm{O}$ (Zemann \& Zemann, 1962; Effenberger, 1977), $\mathrm{CuTeO}_{3}$ (Lindqvist, 1972), $\mathrm{CuTe}_{2} \mathrm{O}_{5}$ (Hanke, Kuprik \& Lindqvist, 1973), $\mathrm{Cu}\left(\mathrm{NH}_{3}\right) \mathrm{TeO}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$ (Johansson \& Lindqvist, 1977). However, information concerning $\mathrm{Cu}^{\mathrm{II}}$ tellurates(VI) is limited to the structure of $\mathrm{Cu}_{3} \mathrm{TeO}_{6}$ (Falck, Lindqvist \& Moret, 1978) which is built up from $\mathrm{Cu}^{2+}$ ions and $\mathrm{TeO}_{6}^{6-}$ ions of $C_{3 i}$ symmetry. In $\mathrm{CuTeO}_{4}$ either a tetrahedral $\mathrm{Te}^{\mathrm{VI}}-\mathrm{O}$ coordination, as has been suggested for $\mathrm{K}_{2} \mathrm{TeO}_{4}$ (Tarte \& Leyder, 1971), or, more probably, a polymerization of $\mathrm{TeO}_{6}$ octahedra, as found in $\mathrm{H}_{2} \mathrm{TeO}_{4}$ (Moret, Philippot, Maurin \& Lindqvist, 1974), is possible.

## Experimental

Single crystals of $\mathrm{CuTeO}_{4}$ were prepared by hydrothermal synthesis. From an aqueous solution of 0.5 M $\mathrm{CuSO}_{4}$ and $0.5 \mathrm{M} \mathrm{Te}(\mathrm{OH})_{6}$, kept at $650^{\circ} \mathrm{C}$ and 2000 bar for one week, single crystals of $\mathrm{Cu}_{3} \mathrm{TeO}_{6}, \mathrm{CuTeO}_{4}$ and $\mathrm{TeO}_{2}$ were obtained simultaneously. The $\mathrm{CuTeO}_{4}$ crystals were very small plates. Slightly larger $\mathrm{CuTeO}_{4}$ crystals could be prepared if the time for the hydrothermal synthesis was extended to two months, and the $\mathrm{Te}(\mathrm{OH})_{6}$ concentration was increased to $1 M$. Some crystallographic data are given in Table 1.

Preliminary investigation with a Weissenberg camera

Table 1. Crystallographic and powder data for $\mathrm{CuTeO}_{4}\left(M D O_{1}\right)$
The cell parameters were refined with the program $P O W D E R$ (Lindqvist \& Wengelin, 1967). The powder data were obtained from a Siemens powder diffractometer.

Crystal habit: yellow plates
Space group: $\mathrm{MDO}_{1}: P 2_{1} / n\left(\mathrm{MDO}_{2}: P \overline{1}\right)$
Plane group symmetry: $\operatorname{Pm}(n) a$ (for $\left[\left.\mathrm{TeO}_{4}\right|_{n} ^{2 n-}\right.$ )

| $\begin{aligned} & a=5.500(2) \AA \\ & b=10.327(7) \end{aligned}$ |  | $V=267.2$ (3) $\dot{\text { A }}^{3}$ |  |
| :---: | :---: | :---: | :---: |
|  |  | $\mu\left(\right.$ Mo Ka) $=193.6 \mathrm{~cm}^{-1}$ |  |
| $c=4.704$ (2) |  | $\rho=6.3$ |  |
| $\beta=90.0$ (1) ${ }^{\circ}$ |  | $M_{r}=255 \cdot 15$ |  |
| $\alpha=\gamma=90^{\circ}$ |  | $Z=4$ |  |
| Powder data |  |  |  |
| hkl | $\sin ^{2} \theta_{0}$ | $\sin ^{2} \theta_{c}$ | $d_{o}(\AA)$ |
| 020 | 0.02231 | 0.02229 | 5.161 |
| 101 | 0.04648 | 0.04650 | $3 \cdot 576$ |
| 111 | 0.05220 | 0.05207 | 3.374 |
| 121 | 0.06883 | 0.06879 | 2.935 |
| 210 | 0.08451 | 0.08415 | 2.652 |
| 040 | 0.08923 | 0.08916 | 2.581 |
| 131 | 0.09662 | 0.09665 | 2.480 |
| 220 | 0.10110 | 0. 10087 | 2.425 |
| 002 | $0 \cdot 10772$ | $0 \cdot 10742$ | 2.349 |
| 041 | $0 \cdot 11619$ | 0.11601 | 2.262 |
| 221 | $0 \cdot 12777$ | $0 \cdot 12772$ | $2 \cdot 157$ |
| 230 | $0 \cdot 12866$ | $0 \cdot 12873$ | $2 \cdot 149$ |
| 022 | 0.12977 | 0. 12971 | $2 \cdot 140$ |
| 112 | 0.13272 | 0. 13264 | 2.116 |
| 240 | 0.16791 | 0.16774 | 1.881 |
| 202 | $0 \cdot 18595$ | $0 \cdot 18600$ | 1.788 |
| 042 | $0 \cdot 19673$ | 0. 19658 | 1.738 |
| 301 | 0.20392 | 0.20366 | 1.707 |
| 311 | 0.20915 | 0.20923 | 1.686 |

showed that $\mathrm{CuTeO}_{4}$ crystallizes with an OD structure (Dornberger-Schiff, 1966), diffuse intensity streaks being observed in the $\mathbf{b}^{*}$ direction. The relative intensities of the streaks were very low compared with the intensities of the discrete reflections. Four different crystals were investigated, but all showed weak intensity streaks only in a few places in reciprocal space. This indicated that the stacking sequence corresponding to an ordered structure is relatively seldom broken.

Reflections corresponding to the unit cell given in Table 1 were measured, since the intensity effect of the continuous streaks could be considered as insignificant. The crystal used for data collection was mounted along [101] on a Pailred two-circle diffractometer. Graphite-monochromated Mo Kor radiation was used, and the intensities were measured with the $\omega$-scan procedure. The scan speed was $2.5^{\circ} \min ^{-1}$ in $\omega$, and the background was measured for 24 s at each end of the scan interval. Nine layer lines were explored out to $2 \theta=60^{\circ}$. Reflections with $h+l=2 n+1$ were very weak, and only 15 such reflections were strong enough to be considered observed. 403 reflections with $\sigma(I) / I<$ 0.3 were regarded as being observed.

The intensities were corrected for Lorentz and polarization effects with the program DATAP1 (Lindgren, 1977) and for absorption with the program DATAP2 (Coppens, Leiserowitz \& Rabinovich, 1965).

## Structure determination

The observed reflections obey the symmetry and extinction rules of $P 2_{1} / n$. The positions of the Te and Cu atoms were obtained from the Patterson function. In the successive electron density calculations, it was possible to find two sets of O atoms that satisfied the intensity data. One such model is shown in Table 2, and the same agreement ( $R=0.048$ ) was obtained if the refinement was carried out with all O atoms shifted by $\left(\frac{1}{2}, 0, \frac{1}{2}\right)$. This indicated that the O atoms have higher symmetry than that inherent in $P 2_{1} / n$. In fact the [ $\mathrm{TeO}_{4}{ }_{n}^{12 n-}$ layer, as refined in $P 2_{1} / n$, very closely obeys the plane group symmetry $\operatorname{Pm}(n) a$. The O atom positions deviate $0.05-0.2 \AA$ in the a direction. The

Table 2. Structure solution refined in space group $P 2_{1} / n$
Equivalent positions: $\pm(x, y, z), \pm\left(\frac{1}{2}+x, \frac{1}{2}-y, \frac{1}{2}, \frac{1}{2}+z\right) ; 15$ positional and 7 thermal parameters varied; $R=\sum| | F_{o}-\left|F_{c}\right| V \sum F_{o}=0.048$.

|  | $x$ | $y$ | $z$ | $B\left(\AA^{2}\right)$ |
| :--- | :---: | :---: | :---: | :---: |
| Te | $0.2963(2)$ | $0.2489(6)$ | $0.0166(2)$ | $0.23(1)$ |
| $\mathrm{Cu}(1)$ | 0 | 0 | 0 | $0.39(5)$ |
| $\mathrm{Cu}(2)$ | $0 . \frac{1}{2}$ | 0 | $\frac{1}{2}(4)$ | $0.55(6)$ |
| $\mathrm{O}(1)$ | $0.537(4)$ | $0.320(2)$ | $0.768(4)$ | $0.4(2)$ |
| $\mathrm{O}(2)$ | $0.535(4)$ | $0.173(2)$ | $0.267(4)$ | $0.7(2)$ |
| $\mathrm{O}(3)$ | $0.271(3)$ | $0.416(1)$ | $0.187(3)$ | $1.0(2)$ |
| $\mathrm{O}(4)$ | $0.271(4)$ | $0.088(2)$ | $-0.164(4)$ | $0.2(2)$ |

geometry of this idealized layer is determined only by the three parameters $y_{\mathrm{O}(01)}, y_{\mathrm{O}(03)}$ and $z_{\mathrm{O}(03)}$ (Table 3). The structure may then be described as built up of such layers stacked according to the translations $\mathrm{L}_{0}, \mathbf{L}_{1}=\mathbf{L}_{0}$ $+\left\lfloor r, \frac{1}{2}, s\right], \mathbf{L}_{2}=\mathbf{L}_{1}+\left[-r, \frac{1}{2},-s\right]=\mathbf{L}_{0}+[0,1,0]$, etc. , where $r$ and $s$ are refinable parameters. The coordinates of the Cu ions are completely determined by the translations (Table 3). The refinement of the constrained structure model (Table 3) gave an $R$ of 0.054 with the least-squares program BLOCK (Lindgren, 1977). Because the number of positional parameters could be reduced from 15 in the $P 2_{1} / n$ model (Table 2) to 5 in the constrained-layer model (Table 3) the parameter standard deviations were improved in spite of the increase in $R$. However, the $R$ difference clearly indicates that the unconstrained $P 2_{1} / n$ model is significantly better than the constrained-layer model, and the detailed description of the structure should thus refer to the $P 2 / n$ coordinates (Table 2).
The scattering factors of Cromer \& Waber (1965) were used for Te and those of Doyle \& Turner (1968) for Cu and O . Correction for the real part of the scattering was applied for Te and Cu (Cromer, 1965). The observations were weighted according to Cruickshank (1970) with $w(h k l)=\left[50 \cdot 0+F_{o}(h k l)+\right.$ $\left.0.01 F_{o}(h k l)^{2}\right]^{-1}$.*

[^0]Table 3. Constrained-layer model of the $\mathrm{CuTeO}_{4}$ structure
The $\left[\mathrm{TeO}_{4}{ }_{n}^{2 h-}\right.$ layers have been constrained to the plane symmetry $P m(n) a$. The Cu atoms only have the ( $n$ ) glide plane in common with the $\left[\mathrm{TeO}_{4}\right]_{n}^{2 n-}$ layers, but their positions are completely determined by the translations $r$ and $s$. The coordinate table is to be extended according to the ( $n$ ) symmetry; equivalent positions: $(x, y, z),\left(\frac{1}{2}+x,-y, \frac{1}{2}+z\right)$. Five positional and four thermal parameters were varied; $R=0.054$.


## Discussion of the structure

Bond distances and angles, both for the unconstrained and the constrained, idealized model, were calculated with DISTAN (written by Zalkin, University of California), and are shown in Table 4.

The idealized layer model is useful for the description of possible disorder in the structure. The layer stacking, as described by the parameters in Table 3, is fully ordered with alternating translations $\left[r, \frac{1}{2} s\right]$ and $\left[-r, \frac{1}{2},-s\right]$. This arrangement thus represents a structure of maximum degree of order (Dornberger-Schiff, 1966), $\mathrm{MDO}_{1}$. The two types of translations are chemically identical with respect to the contacts between the layers and the Cu coordination. It is therefore possible to build up another structure, $\mathrm{MDO}_{2}$,

## $\mathrm{CuTeO}_{4}$



Fig. 1. Schematic picture of the $\mathrm{MDO}_{1}$ and $\mathrm{MDO}_{2}$ structure models showing three adjacent $\left[\mathrm{TeO}_{4}\right)_{n}^{2 n-}$ layers. $\mathrm{MDO}_{2}$ has the unit-cell dimensions $a=5 \cdot 500, b=5.632, c=4.704 \AA, \alpha=$ $90 \cdot 0, \beta=90 \cdot 0, \gamma=66.5^{\circ}$.
in which each translation is positive, i.e. with layers $\mathbf{L}_{0}$, $\mathbf{L}_{1}=\mathbf{L}_{0}+\left[r, \frac{1}{2}, s\right], \mathbf{L}_{2}=\mathbf{L}_{0}+[2 r, 1,2 s]$, etc. Schematic representations of the $\mathrm{MDO}_{1}$ and $\mathrm{MDO}_{2}$ structures are shown in Fig. 1.

Table 4. Interatomic bond distances $(\AA)$ and angles $\left({ }^{\circ}\right)$ in $\mathrm{CuTeO}_{4}$
The columns refer to (a) the unconstrained $P 2_{1} / n$ model and (b) the constrained $P m(n) a$ layer model.

|  |  | (a) | (b) |  |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Te}-\mathrm{O}(4)$ |  | 87 (2) | 1.88 |  |
| $\mathrm{Te}-\mathrm{O}(3)$ |  | 91 (1) | 1.88 |  |
| $\mathrm{Te}-\mathrm{O}(1)$ |  | 91 (2) | 1.97 |  |
| $\mathrm{Te}-\mathrm{O}(2)$ |  | 93 (2) | 1.97 |  |
| $\mathrm{Te}-\mathrm{O}^{\prime}(1)$ |  | 99 (2) | 1.97 |  |
| $\mathrm{Te}-\mathrm{O}^{\prime}(2)$ |  | 02 (2) | 1.97 |  |
| $\mathrm{Cu}(1)-\mathrm{O}(4) \quad 2$ |  | 91 (2) | $2 \times 2.05$ |  |
| $\mathrm{Cu}(1)-\mathrm{O}(3) \quad 2$ |  | 12 (1) | $2 \times 2.17$ |  |
| $\mathrm{Cu}(1)-\mathrm{O}(1) \quad 2$ |  | 26 (2) | $2 \times 2.14$ |  |
| $\mathrm{Cu}(2)-\mathrm{O}(3) \quad 2$ |  | 94 (2) | $2 \times 1.99$ |  |
| $\mathrm{Cu}(2)-\mathrm{O}(2) \quad 2$ |  | 11 (2) | $2 \times 2.22$ |  |
| $\mathrm{Cu}(2)-\mathrm{O}(4) \quad 2$ |  | 22 (2) $2 \times 2.05$ (1) |  |  |
| $\mathrm{O}(1)-\mathrm{Te}-\mathrm{O}^{\prime}(1)$ | 177.7 (1.0) | $\mathrm{O}^{\prime}(1)$ | Te-O(3) | 91.2 (9) |
| $\mathrm{O}(1)-\mathrm{Te}-\mathrm{O}(2)$ | $93 \cdot 3$ (9) | $\mathrm{O}^{\prime}(1)$ | Te-O(4) | 84.2 (9) |
| $\mathrm{O}(1)-\mathrm{Te}-\mathrm{O}^{\prime}(2)$ | 89.1 (8) | $\mathrm{O}(2)$ | $\mathrm{Te}-\mathrm{O}^{(2)}$ | 177.5 (1-2) |
| $\mathrm{O}(1)-\mathrm{Te}-\mathrm{O}(3)$ | 87.7 (8) | $\mathrm{O}(2)$ | $\mathrm{Te}-\mathrm{O}(3)$ | 99.2 (8) |
| $\mathrm{O}(1)-\mathrm{Te}-\mathrm{O}(4)$ | 86.5 (9) | $\mathrm{O}(2)$ | $\mathrm{Te}-\mathrm{O}(4)$ | 88.1 (9) |
| $\mathrm{O}^{\prime}(1)-\mathrm{Te}-\mathrm{O}(2)$ | 88.8 (9) | $\mathrm{O}^{\prime}(2)$ | - $\mathrm{e}-\mathrm{O}(3)$ | 92.2 (9) |
| $\mathrm{O}^{\prime}(1)-\mathrm{Te}-\mathrm{O}^{\prime}(2)$ | 88.8 (9) | $\mathrm{O}(3)$ | $\mathrm{Te}-\mathrm{O}(4)$ | 171.3 (8) |
| $\mathrm{O}(1)-\mathrm{Cu}(1)-\mathrm{O}(3)$ |  | 96.0 ( | 6), $85 \cdot 0$ |  |
| $\mathrm{O}(1)-\mathrm{Cu}(1)-\mathrm{O}(4)$ |  | 103.7 | 8), $76 \cdot 3$ |  |
| $\mathrm{O}(3)-\mathrm{Cu}(1)-\mathrm{O}(4)$ |  | 90.7 | 7), $89 \cdot 3$ |  |
| $\mathrm{O}(2)-\mathrm{Cu}(2)-\mathrm{O}(3)$ |  | 102.4 | 7), $77 \cdot 6$ |  |
| $\mathrm{O}(2)-\mathrm{Cu}(2)-\mathrm{O}(4)$ |  | 94.3 | 8), 85.7 |  |
| $\mathrm{O}(3)-\mathrm{Cu}(2)-\mathrm{O}(4)$ |  | 94.0 | 7), $86 \cdot 0$ |  |

Table 5. Copper coordination in tellurates

| Compound | Type | $\begin{aligned} & \text { Bond distances }(\AA) \\ & \mathrm{Cu}-\mathrm{O} \end{aligned}$ | $\sigma(\AA)$ | Reference |
| :---: | :---: | :---: | :---: | :---: |
| (a) $\mathrm{Te}(\mathrm{IV})$ |  |  |  |  |
| CuTeO $3.2 \mathrm{H}_{2} \mathrm{O}$ | Square planar | $\begin{aligned} & 1 \cdot 93,1 \cdot 95,1.96 \\ & 1.97,2 \cdot 35,3.37 \end{aligned}$ | 0.01 | 1 |
| $\mathrm{CuTeO}_{3}$ | Irregular | $\begin{aligned} & 1.94,1.97,1.97 \\ & 1.98,2.38,3.33 \end{aligned}$ | 0.02 | 2 |
| $\mathrm{CuTeO}_{3}$ (h.p.) | Square planar (distorted octahedral) | $\begin{aligned} & 2 \times 1.91,2 \times 2.06, \\ & 2 \times 2.61 \end{aligned}$ | 0.02 | 3 |
| $\mathrm{Cu}\left(\mathrm{NH}_{3}\right) \mathrm{TeO}_{3} \cdot \mathrm{H}_{2} \mathrm{O}$ | Square planar | $\begin{aligned} & 1.950,1 \cdot 958,1 \cdot 960 \\ & 1.998, * 2 \cdot 602,3.066 \end{aligned}$ | 0.008 | 4 |
| $\mathrm{CuTe} \mathrm{O}_{5}$ | Square planar (distorted octahedral) | $\begin{aligned} & 1.950,1.952,1.961 \\ & 1.969,2.305,2.780 \end{aligned}$ | 0.003 | 5 |
| (b) Te (VI) |  |  |  |  |
| $\mathrm{Cu}_{3} \mathrm{TeO}_{6}$ | Square planar (distorted octahedral) | $\begin{aligned} & 2 \times 1.949,2 \times 2.031, \\ & 2 \times 2.369 \end{aligned}$ | 0.002 | 6 |
| $\mathrm{CuTeO}_{4}$ | Octahedral | cf. Table 4 | 0.01 | 7 |

References: (1) Effenberger (1977). (2) Lindqvist (1972). (3) Philippot \& Maurin (1976). (4) Johansson \& Lindqvist (1977). (5) Hanke, Kupčik \& Lindqvist (1973). (6) Falck, Lindqvist \& Moret (1978). (7) This work.

* $\mathrm{Cu}-\mathrm{N}$ bond distance.

The reciprocal lattices of the $\mathrm{MDO}_{1}$ and $\mathrm{MDO}_{2}$ structures coincide in the $h 0 l$ plane, but the contribution of the $\mathrm{MDO}_{2}$ structure in the crystal used for the structure determination was too small to appear as a significant increase in the $F_{o} / F_{c}$ ratio for the $h 0 l$ reflections, compared with the $h k l$ reflections. Even in the other crystals investigated, the $\mathrm{MDO}_{2}$ arrangement did not seem to occur to more than a few percent, although weak $\mathrm{MDO}_{2}$ reflections could be detected from one of the crystals.

It should be noted that the refinement in $P 2_{1} / n$ shows that the O atoms deviate slightly ( $0.05-0.15 \AA$ ) from the mirror plane in the plane group symmetry $\operatorname{Pm}(n) a$, and $x_{\mathrm{O}(01)}=x_{\mathrm{O}(02)}=0.241$ and $x_{\mathrm{O}(03)}=x_{\mathrm{O}(04)}=$ -0.025 , instead of $\frac{1}{4}$ and 0 respectively. In addition, the $\left[r, \frac{1}{2}, s\right]$ translation does not hold strictly (in the $P 2_{1} / n$ refinement results) for the $x$ coordinates of the O atoms. However, the appearance of diffuse streaks in the $\mathbf{b}^{*}$ direction confirms the OD character of the investigated crystals.

The results may be interpreted as indicating that the Cu coordination of O atoms tends to distort the $\left[\mathrm{TeO}_{4}{ }_{n}{ }_{n}^{2 n-}\right.$ layers in a way which is more easily achieved in the $\mathrm{MDO}_{1}$ than in the $\mathrm{MDO}_{2}$ type of stacking sequence. In a related structure under investigation, $\mathrm{PbTeO}_{4}$, similar disordered stacking occurs, but the two MDO structures occur to approximately the same extent.

The following discussion of the $\mathrm{CuTeO}_{4}$ structural details refers to the unconstrained $P 2_{1} / n$ model (Table 2).

Octahedral layers of composition $\left[\mathrm{TeO}_{4}\right]_{n}^{2 n-}$ have previously been found in $\mathrm{Te}_{2} \mathrm{O}_{5}$ (Lindqvist \& Moret, 1973) and in $\mathrm{H}_{2} \mathrm{TeO}_{4}$ (Moret, Philippot, Maurin \& Lindqvist, 1974). Both these structures are fully ordered and belong to the monoclinic space groups $P 2_{1}$ and $P 2_{1} / n$ respectively. The layers found in $\mathrm{CuTeO}_{4}$ and $\mathrm{Te}_{2} \mathrm{O}_{5}$ can schematically be described in the same way (Fig. 2). The $\mathrm{Te}-\mathrm{O}_{\text {bridge }}$ bonds in $\mathrm{CuTeO}_{4}$ are slightly longer than those in $\mathrm{H}_{2} \mathrm{TeO}_{4}$ and $\mathrm{Te}_{2} \mathrm{O}_{5}$, which


Fig. 2. A comparison of the $\left[\mathrm{TeO}_{4}\right]_{n}^{2 n-}$ layers found in $\mathrm{H}_{2} \mathrm{TeO}_{4}$, $\mathrm{CuTeO}_{4}$ and $\mathrm{Te}_{2} \mathrm{O}_{5}$.
have average values of 1.918 and $1.935 \AA$ respectively. The $\mathrm{Te}-\mathrm{O}_{\text {terminal }}$ bonds of $1.87-1.91 \AA$ in $\mathrm{CuTeO}_{4}$ are in good agreement with values (1.83-1.91 $\AA$ ) found in other tellurates(VI) (Lindquist, 1973). In $\mathrm{Te}_{2} \mathrm{O}_{5}$ there are two different $\mathrm{Te}-\mathrm{O}_{\text {terminal }}$ bonds of 1.848 and $1.929 \AA$, as a result of the different connections with the interlayer $\mathrm{Te}(\mathrm{IV})$ atom.

The Cu coordination in $\mathrm{CuTeO}_{4}$ has been assumed to cause slight distortion of the $\left[\mathrm{TeO}_{4}\right]_{n}^{]^{n-}}$ layers. However, the packing of these layers mainly determines the geometry of the $\mathrm{CuO}_{6}$ polyhedron, which is a distorted octahedron (Table 4). The Cu coordination in different tellurates is summarized in Table 5.

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[^0]:    * A list of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. SUP 33292 ( 4 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CHI 1NZ, England.

